

**Example 9.1**

$\text{K}_2\text{SO}_4$  ( $M=174$ ) has a solubility in water @  $20^\circ\text{C}$  of  $109 \text{ g K}_2\text{SO}_4/\text{kg water}$ , a solution which has a density of  $1.080 \text{ g/cm}^3$ . A supersaturated solution of  $116 \text{ g K}_2\text{SO}_4/\text{kg water}$  has a density of  $1.090 \text{ g/cm}^3$ . Find values for  $\Delta c$ ,  $S$ ,  $\sigma$  on the 5 different bases.