

Alcohol Dehydrogenase

April 7, 2005

Summary

Alcohol dehydrogenase catalyzes the oxidation of ethanol to acetaldehyde using NAD as a co-substrate. This protocol describes a direct enzyme assay for determining alcohol dehydrogenase activity.

Solutions Required

1. 1.0 M tris-HCl buffer pH = 8.8:
adjust to a pH of 8.8 with 20% KOH.
2. 10 mM potassium phosphate buffer pH = 7.0
prepared by mixing 8 mL of 10 mM KH_2PO_4 and 20 mL of 10 mM K_2HPO_4
or prepared by dissolving 0.0194 g KH_2PO_4 and 0.0815 g $\text{K}_2\text{HPO}_4 \cdot 3\text{H}_2\text{O}$ in 50 mL water.
3. ethanol
4. 4.0 mM NAD^+
must be prepared fresh

Preparation of Cell Extract

Follow general protocol **Preparation of Cell Extract**.

1. After first pelletization of cells, resuspend at 4°C in potassium phosphate buffer.
2. After second pelletization of cells, resuspend at 4°C in potassium phosphate buffer.

Spectrophotometer

Turn on the ultraviolet bulb on the spectrophotometer (Beckman DU50) and wait 10 minutes for warm-up. Select the kinetics-time window on the instrument. Load the method "A:/nadh30" or "A:/nadh37". These methods each have a run-time of 60 s, a temperature of 30°C or 37°C (respectively), a wavelength of 340 nm and use 2 autosamplers.

Procedure

1. For each assay, prepare the two cocktails shown in the following table into two separate UV-translucent cuvettes, and keep them on ice.

Solution	<u>Volume (μL) added to:</u>	
	Control	Experimental
DI H ₂ O	600	500
1.0 M tris-HCl	250	250
NAD ⁺	100	100
ethanol	0	100

2. Directly from the ice when ready to commence the assay, place the two quartz cuvettes (each containing 900 μL) into the spectrophotometer holder (position #1 for control, position #2 for experimental). Use cuvette lid caps to mix 3 or 4 times then insert in instrument.
3. Wait 10 minutes to allow the temperature of the solutions in the cuvettes to equilibrate.
4. "Blank" and then depress "Read Samples" on the monitor.
5. Simultaneously add 50 μL † of the cell extract to the cuvettes.
6. To mix solutions, immediately and simultaneously aspirate and dispense the contents of the cuvettes with a pipettor. Mix the solutions in this way ten times. (Count!)
7. Promptly depress "start" on the monitor.
8. Record the rates for the two (control and experimental) cuvettes.

† Dilution of the cell extract may be adjusted so that change in absorbance is between about 0.05 and 0.7 AU in one minute. This dilution should be accomplished externally in a microcentrifuge tube (for example, by adding 50 μL of cell extract to 950 μL DI water to achieve a dilution of 20). The volume of 50 μL should always be used in the enzyme assay mixture.

Calculation of Activity

One unit (U) of alanine dehydrogenase activity is defined as the amount of enzyme required to produce 1.0 μmole of pyruvate in one minute.

1.
$$dA/dt (\text{min}^{-1}) = [\text{Rate}]_{\text{experimental}} - [\text{Rate}]_{\text{control}} = dA/dt$$

2.
$$\text{Activity} = \frac{1000 \times TV \times D \times dA/dt}{\epsilon \times V \times CF}$$

Activity: Volumetric Activity (U/L)

TV: Total volume in cuvette (1000 μL)

D: Dilution of the cell extract. (For example, if 50 μL of cell extract were added to 950 μL DI water prior to using a volume of cell extract in the assay, then D=20)

V: Volume of cell extract used (50 μL)

ϵ : Molar extinction coefficient for NADH (6.22 L/mmol for a path length of 1.0 cm)

CF: Concentration Factor of cell extract (For example, if a 100 mL sample is concentrated to a 2 mL volume for the French Press, then CF=50)

$$3. \quad \text{Specific Activity} = \frac{\text{Activity}}{\text{Protein Concentration}} \cdot 1$$

Activity: Volumetric Activity, as calculated in #2 above (U/L)

Protein Concentration: Protein concentration, as calculated in protocol **Total Protein Concentration** (mg/L)

Specific Activity: (U/mg protein)

Reference

A. Blandino, I. Caro, D. Cantero (1997) "Comparative Study of Alcohol Dehydrogenase Activity in Flor Yeast Extracts, Biotechnology Letters, 19(7), 651-654.